



Thickness-controllable synthesis of metal-organic framework based hollow nanoflowers with magnetic core *via* liquid phase epitaxy for phosphopeptides enrichment

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ABSTRACT

A thickness-controllable method for preparing metal-organic framework hollow nanoflowers on magnetic cores ($\text{Fe}_3\text{O}_4@$ MOFs HF) was demonstrated for the first time. The petal of magnetic core with hollow nanoflower structure served as medium for assembling UiO-66-NH₂ shell with different thickness. To further improve its performance, Zr⁴⁺ was immobilized on the surface of $\text{Fe}_3\text{O}_4@$ UiO-66-NH₂. Compared with conventional $\text{Fe}_3\text{O}_4@$ UiO-66-NH₂-Zr⁴⁺ nanospheres, the $\text{Fe}_3\text{O}_4@$ UiO-66-NH₂-Zr⁴⁺ HF showed increased enrichment performance for phosphopeptides. The $\text{Fe}_3\text{O}_4@$ UiO-66-NH₂-Zr⁴⁺ HF served as an attractive restricted-access adsorption material exhibited good selectivity ($m_{\beta\text{-casein}}:m_{\text{BSA}}=1:1000$), high sensitivity (1.0 fmol) and excellent size-exclusion effect ($m_{\beta\text{-casein digests}}:m_{\text{BSA}}=1:200$). Furthermore, the $\text{Fe}_3\text{O}_4@$ UiO-66-NH₂-Zr⁴⁺ HF was successfully applied to the specific capture of ultratrace phosphopeptide from complex biological samples, revealing the great potential for the identification and analysis of trace phosphopeptides in clinical analysis. This work can be easily extended to the fabrication of diverse mag-MOF HF with multifunctional and easy to post-modify properties, and open up a new avenue for the design and construction of new MOFs material.

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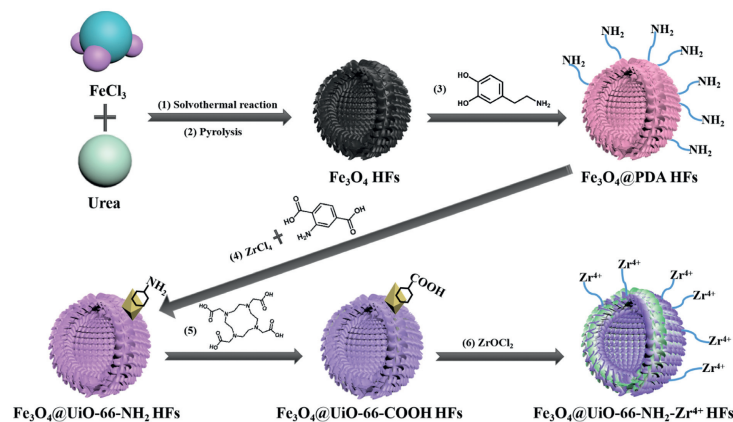
Heterogeneous nanomaterials are considered as one of the most important candidates in widespread applications [1-5]. Considerable attention has been focused on the heterostructure from porous organic framework materials to prepare alloyed, doped, or hybrid nanomaterials for enhanced adsorption or selective performance [6-8]. Metal organic frameworks (MOFs) are connected *via* organic linkers and metal ions/clusters to form porous frameworks, and have attracted much attention [9-11]. MOFs provide an extensive combination of building units (inorganic nodes and organic linkers) to achieve desired structures and functionalities [12]. Therefore, many efforts have been devoted to prepare MOF-based heterogeneous nanomaterials for enhancing the monomeric performance. Regardless of doping any materials, the new structures and novel properties are generated in the hybrid heterogeneous nano-

materials [13-16]. A growing number of hybrid MOF-based heterogeneous nanomaterials have been prepared and used in various fields.

Previous studies on hybrid MOF-based heterogeneous nanomaterials mainly focused on relatively single core-shell shape, which led to the inability to maximize properties of heterogeneous structures [2,4,17,18]. As one of the most important parameters, the shape of materials usually has significant effect on their performance. The hollow MOFs were successfully fabricated and showed boosted surface area, superb porosity, and excellent pore accessibility, and exhibited a significantly improved performance [19]. Wang *et al.* reported the dendritic Ag nanocrystals mimicking the structural feature (dendritic) of moth's antennae. The existence of numerous cavity traps in Ag dendritic nanocrystals prolonged reaction time of the gaseous molecules on the surface of solid surface through the "cavity-vortex" effect bring the more sensitive detection sensitivity [20]. A large number of works have verified the positive influence of the shape on material performance [19-22]. But, no works has thus far paid close attention in heteroge-

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Scheme 1. The formation processes of $\text{Fe}_3\text{O}_4@UiO-66-NH_2-Zr^{4+}$ HF.

neous nanomaterials with the special shapes, which hindered the exploration of controllable design of novel heterogeneous materials complexes.

Thus, the aim of this work is to provide a viable and versatile strategy for preparing thickness-controllable MOF based hollow nanoflowers with magnetic core ($\text{Fe}_3\text{O}_4@MOFs$ HF). Here, the magnetic core hollow nanoflowers (Fe_3O_4 HF) were firstly synthesized. The Fe_3O_4 HF were employed as the core, metal organic frameworks shell with magnetic core ($\text{Fe}_3\text{O}_4@MOFs$ HF) was prepared by liquid phase epitaxy. The thickness of the MOFs shell was randomly controlled by changing the period of assembly. To achieve higher performance, $\text{Fe}_3\text{O}_4@UiO-66-NH_2$ HF were post-modified with Zr^{4+} ion to generate $\text{Fe}_3\text{O}_4@UiO-66-NH_2-Zr^{4+}$ HF, which was successfully used to separate and enrich low-abundance phosphopeptides from complex biological samples. This work provides a new avenue for the design and construction of hybrid MOF-based heterogeneous nanomaterials with hollow nanoflower morphology.

The schematic design of the $\text{Fe}_3\text{O}_4@UiO-66-NH_2-Zr^{4+}$ HF was shown in Scheme 1. Firstly, the precursor hollow nanoflowers (PC HF) were obtained by the solvothermal reaction with FeCl_3 and urea, and then transformed into Fe_3O_4 HF after calcination. The polydopamine (PDA) was modified on the surface of Fe_3O_4 HF, and the NH_2 groups of PDA could react with metal ions ($\text{Fe}_3\text{O}_4@PDA$ HF). Then, $\text{Fe}_3\text{O}_4@UiO-66-NH_2$ HF was prepared by the assembling of $\text{Fe}_3\text{O}_4@PDA$ HF and MOF stock solution. Finally, the MOF surface was functionalized by $-COOH$, and Zr^{4+} was then immobilized on the material by $-COOH$. During the process of Zr^{4+} immobilization, the coordination interaction between $-COOH$ group and Zr^{4+} served as the driving force. Surprisingly, the thickness of the MOFs shell could be controlled by changing the period of assembly.

The structures and morphologies of the prepared materials were thoroughly characterized by scanning electron microscopy (SEM) and transmission electron microscopy (TEM). It was clear that the hollow nanoflowers was organized from nanosheets as building blocks (Figs. 1A, E and I). After a mild heat-treatment was applied on PC HF, the Fe_3O_4 with thinner and folded nanosheets was obtained (Figs. 1B, F and J). After stirring with PDA, the edges of magnetic nanoflowers were wrapped by the thin layers, which indicated that the PDA was firmly covered on the surface of Fe_3O_4 HF (Figs. 1C, G and K). The surface of nanopetals were then assembled by a layer of MOF after heated with MOF stock solution for a period of time (Figs. 1D, H and L). From Fig. 1D and Fig. S1 (Supporting information), as the increase of assembly cycles, the thickness of the nanosheets increased until the space between the petals were fully filled up, and then a pure MOF shells were

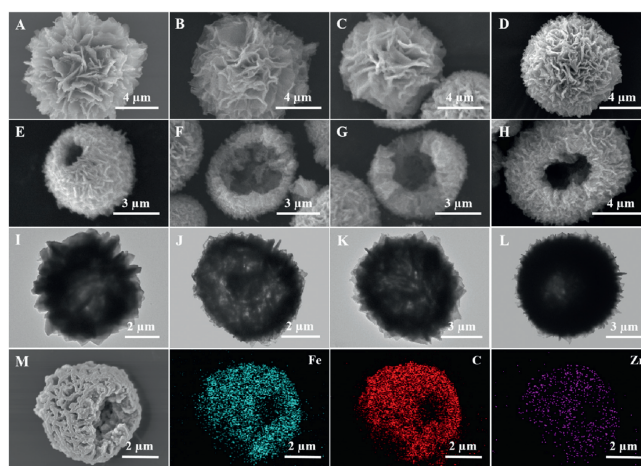


Fig. 1. SEM images of the PC HF (A), Fe_3O_4 HF (B), $\text{Fe}_3\text{O}_4@PDA$ HF (C), $\text{Fe}_3\text{O}_4@UiO-66-NH_2$ HF (D) with different magnifications. SEM and TEM images for inner hollow structure of PC HF (E, I), Fe_3O_4 HF (F, J), $\text{Fe}_3\text{O}_4@PDA$ HF (G, K), $\text{Fe}_3\text{O}_4@UiO-66-NH_2$ HF (H, L). (M) Elemental mapping of the $\text{Fe}_3\text{O}_4@UiO-66-NH_2$ HF.

formed on the surface of Fe_3O_4 HF. Comparing Fig. 1C and Fig. S1A, after one cycle, the thickness of the MOF layer was approximately 80 nm. After another cycle, the thickness of the MOF layer further increased by approximately 163 nm (Fig. S1B). The material that had undergone 5 cycles was completely covered by MOF layer (Figs. S1C and D). Furthermore, the distributions of elements from the area mapping showed a uniform distribution of the metallic distribution of Fe, C and Zr in the $\text{Fe}_3\text{O}_4@UiO-66-NH_2$ HF, which highly indicated that the UiO-66- NH_2 nanoparticles were formed on the surface of Fe_3O_4 HF and the $\text{Fe}_3\text{O}_4@UiO-66-NH_2$ products still kept the hollow structure (Fig. 1M).

Fourier transform infrared spectra (FT-IR) of materials were presented in Fig. 2A. Compared with PC, other HF materials possessed the characteristic peaks (546 cm^{-1}) belonging to Fe-O [23]. Compared with Fe_3O_4 HF, the new peak appeared at around 1490 cm^{-1} in $\text{Fe}_3\text{O}_4@PDA$ HF, which attributed to the aromatic rings of PDA shell [24]. The FT-IR spectrum of $\text{Fe}_3\text{O}_4@UiO-66-NH_2$ obtained after the assembling between the $\text{Fe}_3\text{O}_4@PDA$ HF and reaction solution of UiO-66- NH_2 . The characteristic peaks appeared at $1651, 1559, 1434, 1385\text{ cm}^{-1}$ could be ascribed to carboxyl groups stretching modes in the $\text{Fe}_3\text{O}_4@UiO-66-NH_2$ [25]. The absorption bands of 1250 and 764 cm^{-1} were referred to the C-N and N-H stretching bands in the 2-aminoterephthalic acid, confirming the existence of $-NH_2$ groups [26].

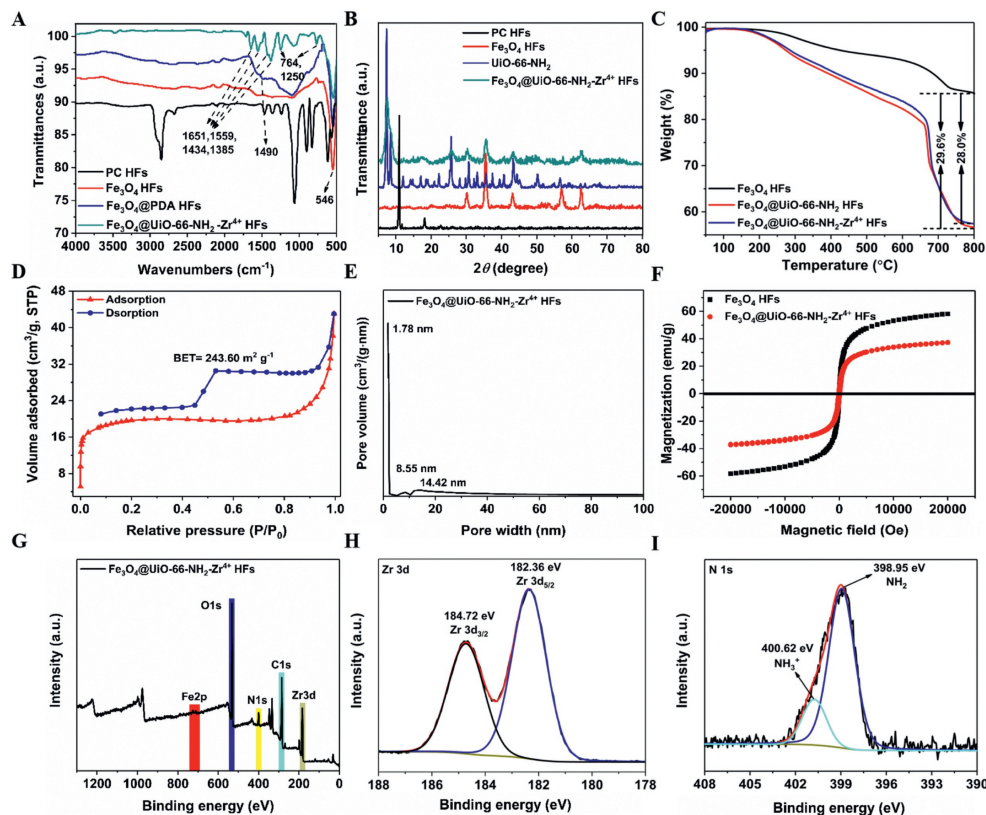


Fig. 2. (A) FTIR spectra of PC, Fe_3O_4 , Fe_3O_4 @PDA, Fe_3O_4 @UiO-66- NH_2 . (B) XRD patterns of PC, Fe_3O_4 , UiO-66- NH_2 , Fe_3O_4 @UiO-66- NH_2 - Zr^{4+} . (C) TGA curves of Fe_3O_4 , Fe_3O_4 @UiO-66- NH_2 , Fe_3O_4 @UiO-66- NH_2 - Zr^{4+} . Nitrogen adsorption/desorption isotherms (D) and pore radial distribution (E) of Fe_3O_4 @UiO-66- NH_2 - Zr^{4+} . (F) Magnetic curves of Fe_3O_4 HFs and Fe_3O_4 @UiO-66- NH_2 - Zr^{4+} . XPS spectra of Fe_3O_4 @UiO-66- NH_2 - Zr^{4+} HFs (G), Zr 3d (H) and N 1s (I).

In order to further verify the existence of MOF shell and whether post-modification would affect the crystallinity of MOF, X-ray diffractometer (XRD) of PC, Fe_3O_4 , UiO-66- NH_2 and Fe_3O_4 @UiO-66- NH_2 - Zr^{4+} HFs were further studied. Compared the spectra of PC and Fe_3O_4 in Fig. 2B, the typical diffraction peaks of Fe_3O_4 (30.2° (220), 35.3° (311), 43.3° (400), 57.1° (511), 62.6° (440)) appeared newly in the Fe_3O_4 spectrum, which indicated the properties of nanomaterials were changed from non-magnetic to magnetic during the high-temperature calcination process. From the spectra of Fe_3O_4 , UiO-66- NH_2 and Fe_3O_4 @UiO-66- NH_2 - Zr^{4+} HFs, the additional new diffraction peaks at around 7.5° and 25.8° could be assigned to the crystalline structure of the MOFs (UiO-66- NH_2).

To examine the structural formation, the thermal gravimetric analyser (TGA) of materials were studied (Fig. 2C). Compared with Fe_3O_4 , the Fe_3O_4 @UiO-66- NH_2 HFs showed extra weight loss of almost 28.0%, indicating that the MOF shell was high yielding. Meanwhile, the TGA analysis of the Fe_3O_4 @UiO-66- NH_2 - Zr^{4+} HFs showed the extra weight loss which up to 29.6% compared with the Fe_3O_4 . The calculated weight loss of 1.6% was attributed to the post-synthetic modifications of Zr^{4+} , also indicating that the Zr^{4+} groups was high yielding. The large loading yields of MOF and Zr^{4+} would help increase performance of the as-prepared materials. The surface area and pore structure were investigated. As shown in Fig. 2D, the Fe_3O_4 @UiO-66- NH_2 - Zr^{4+} HFs composites displayed typical IV sorption isotherm profiles with the H4 hysteresis loop at higher pressures that revealed the existence of slit-like mesopore [27]. From Figs. 2D and E, the Brunauer-Emmett-Teller (BET) surface areas and pore size of Fe_3O_4 @UiO-66- NH_2 - Zr^{4+} HFs were determined to be $243.60 \text{ m}^2/\text{g}$ and 1.78, 8.55, 14.42 nm, respectively.

The magnetic curves of the Fe_3O_4 @UiO-66- NH_2 - Zr^{4+} HFs and Fe_3O_4 HFs were measured. As shown in Fig. 2F, the saturation

magnetization values of Fe_3O_4 @UiO-66- NH_2 - Zr^{4+} HFs and Fe_3O_4 HFs were approximately 31.3 and 58.2 emu/g indicating that the surface UiO-66- NH_2 - Zr^{4+} modification decreased the magnetic properties of original Fe_3O_4 HFs. To obtain detailed elemental compositions and electronic states, X-ray photoelectron spectroscopy (XPS) analysis was further carried out. From Fig. 2G, the survey spectrum of Fe_3O_4 HFs and Fe_3O_4 @UiO-66- NH_2 - Zr^{4+} HFs showed the presence of Fe 2p, C 1s, O 1s and Fe 2p, Zr 3d, C 1s, N 1s, O 1s elements, respectively. As seen from Fig. 2H, the binding peaks at around 182.36 and 184.72 eV were attributed to Zr $3d_{5/2}$ and Zr $3d_{3/2}$, respectively [28]. The binding energy of N 1s emission peak (Fig. 2I) observed at 398.95 and 400.62 eV, which were ascribed to the $-\text{NH}_2$ and $-\text{NH}_3^+$, respectively. Therefore, the above results obviously confirmed that the formation of Fe_3O_4 @UiO-66- NH_2 - Zr^{4+} HFs.

To demonstrate the feasibility of Fe_3O_4 @UiO-66- NH_2 - Zr^{4+} HFs as the affinity platform for enrichment of phosphopeptides, the tryptic digest of the standard phosphoprotein (β -casein) was employed as the testing sample (Fig. 3). Without any pretreatment procedure, the signals were dominated by nonphosphopeptides and only very weak signal intensity of phosphopeptide (β 1s) was detected due to the low abundance and serious signal suppression by the nonphosphopeptides (Fig. 3A). Although the signals of phosphopeptide was enhanced after treatment by Fe_3O_4 @UiO-66- NH_2 HFs, a number of non-phosphopeptide peaks could still be observed. Otherwise, the relatively high signal-to-noise ratio of the baseline was presented (Fig. 3B). In contrast, after enrichment with Fe_3O_4 @UiO-66- NH_2 - Zr^{4+} HFs, strong signals of phosphopeptide dominated the spectrum and the clean mass spectrometry (MS) background was obtained (Fig. 3C). In addition, three phosphopeptides (β 1s, β 2s and β 3m) and their dephosphopeptide

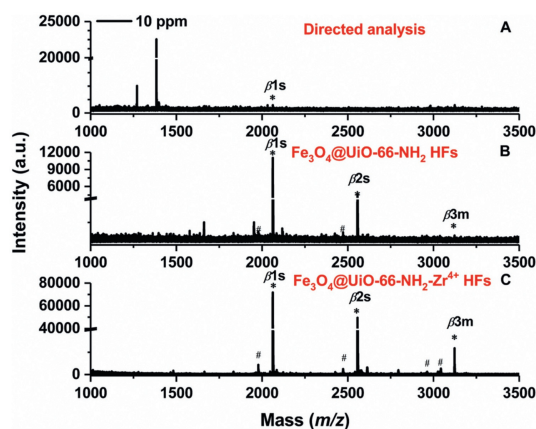


Fig. 3. MALDI-TOF mass spectra of the tryptic digests of β -casein: direct analysis (A), analysis after enrichment with $\text{Fe}_3\text{O}_4@UiO-66-NH_2$ HF (B), analysis after enrichment with $\text{Fe}_3\text{O}_4@UiO-66-NH_2-Zr^{4+}$ HF (C). “*” and “#” indicate phosphorylated peptides and their dephosphorylated counterparts, respectively.

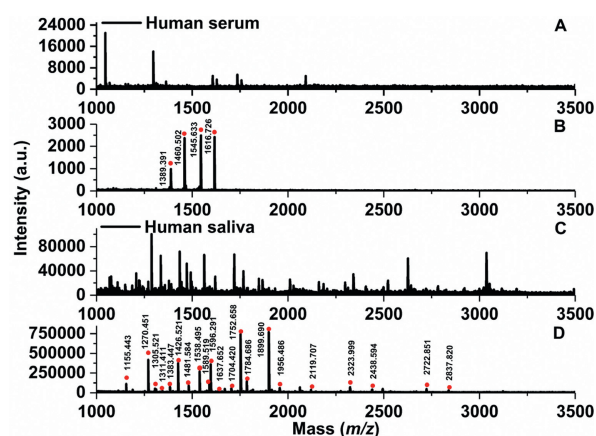


Fig. 4. MALDI-TOF mass spectra of the tryptic digests of human serum and saliva: Direct analysis (A, C) and analysis after enrichment by the $\text{Fe}_3\text{O}_4@UiO-66-NH_2-Zr^{4+}$ HF (B, D).

were all detected with strong signals. The details of the phosphopeptides were listed in Table S1 (Supporting information).

To prove that hollow nanoflower structure helped to improve extraction performance, the extraction capacity of $\text{Fe}_3\text{O}_4@UiO-66-NH_2-Zr^{4+}$ HF and $\text{Fe}_3\text{O}_4@UiO-66-NH_2-Zr^{4+}$ NPs for phosphopeptides were compared under the same conditions. From Fig. S2 (Supporting information), after enrichment with $\text{Fe}_3\text{O}_4@UiO-66-NH_2-Zr^{4+}$ HF, the intensity of the three characteristic signal peaks of the phosphopeptide ($\beta 1s$, $\beta 2s$, $\beta 3m$) were all much higher than that after enrichment with $\text{Fe}_3\text{O}_4@UiO-66-NH_2-Zr^{4+}$ NPs. $\text{Fe}_3\text{O}_4@UiO-66-NH_2-Zr^{4+}$ HF showed better enrichment performance might be attributed to the following reasons. Firstly, the hierarchical pores of the special nanoflower structure could more freely allow targets to enter the surface and cavity, providing more accessible sites. Simultaneously, the hollow structure made the active sites inside the material more accessible, ensuring the material properties was used to the greatest extent. Finally, the cavity could hold more targets, accompanied by an increase in adsorption efficiency [29]. The conclusion was summarized that the unique shape of hollow nanoflowers played a crucial role in the selective enrichment of phosphopeptides.

To investigate selectivity of $\text{Fe}_3\text{O}_4@UiO-66-NH_2-Zr^{4+}$ HF toward phosphopeptides, the tryptic digest mixture of β -casein and bovine serum albumin (BSA) at different molar ratios (1:1000) were used as imitative biological sample. Before enrichment, abundant nonphosphopeptides peaks dominated the chromatogram accompanying with relatively high baseline signal. No any peaks of phosphopeptides were detected, which could be caused by the MS signals of phosphopeptides in the mixture solution (β -casein:BSA=1:1000) were seriously suppressed by the nonphosphopeptides from BSA digest (Fig. S3A in Supporting information). However, after treatment with $\text{Fe}_3\text{O}_4@UiO-66-NH_2-Zr^{4+}$ HF, three phosphopeptides ($m/z = 2061.9$, 2566.9 and 3122.6) with high intensities and a relatively clear background were achieved (Fig. S3B in Supporting information), which was better than previously reported materials of MCNC@COF@Zr⁴⁺ (1:200) [30], mSiO₂@PEI (1:100) [31], SiO₂@ATP-Ti⁴⁺ (1:1) [32], $\text{Fe}_3\text{O}_4@C$ -phosphate-Zr⁴⁺ (1:25) [33], and $\text{Fe}_3\text{O}_4@nSiO_2@mSiO_2/TiO_2-Ti^{4+}$ (1:50) [34]. The results indicated the high selectivity of $\text{Fe}_3\text{O}_4@UiO-66-NH_2-Zr^{4+}$ HF as affinity platform for enrichment of phosphopeptides.

Considering the special hollow nanoflower structure of material, it was expected that $\text{Fe}_3\text{O}_4@UiO-66-NH_2-Zr^{4+}$ HF could exclude the large-size interfering substances in the enrichment of phosphopeptides from complex mixture. To test size-exclusion ef-

fect, the β -casein digest was mixed with BSA directly at a molar ratio of 1:200 was chosen as a model sample. From Fig. S4C (Supporting information), BSA protein was obviously observed in high molecular weight region without enrichment, while no phosphopeptides could be detected in the low molecular weight region and abundant nonphosphopeptides peaks dominated the chromatogram (Fig. S4A in Supporting information). On the contrary, after treatment by $\text{Fe}_3\text{O}_4@UiO-66-NH_2-Zr^{4+}$ HF, three phosphopeptides with strong intensity were identified (Fig. S4B in Supporting information) and no signal of BSA protein was observed (Fig. S4D in Supporting information). The result indicated that the affinity of $\text{Fe}_3\text{O}_4@UiO-66-NH_2-Zr^{4+}$ HF toward small-size phosphopeptides would be not affected by the large proteins, indicating its potential application in complex biosamples.

To further explore the sensitivity of the proposed method for the enrichment of phosphopeptides, tryptic digests of β -casein with different concentrations were investigated based on the $\text{Fe}_3\text{O}_4@UiO-66-NH_2-Zr^{4+}$ HF (Fig. S5 in Supporting information). Three phosphopeptides were still clearly detected with a relatively high signal (signal-to-noise ratio > 3) even at the concentration as low as 10 fmol (Fig. S5B). Surprisingly, one phosphopeptide peak ($\beta 1s$) could still be identified when the concentration was as low as 1 fmol (Fig. S5C). The result showed that the prepared material exhibited the better performance than previously reported [30,35–37]. Undoubtedly, $\text{Fe}_3\text{O}_4@UiO-66-NH_2-Zr^{4+}$ HF had good affinity for phosphopeptides.

Encouraged by all achieved results, $\text{Fe}_3\text{O}_4@UiO-66-NH_2-Zr^{4+}$ HF were further applied to capture endogenous phosphopeptide in complex biosamples of human saliva and serum. we had obtained an approval from the Ethics Committee of Biomedical Scientific Research Henan University (HUSOM2024-394). As a widely used clinical sample, human saliva and serum were easy to get and contained many low-abundance endogenous phosphopeptides which could be the potential biomarkers for disease diagnosis. For human serum, without enrichment, nonphosphopeptides dominated the spectrum and no any signal of phosphopeptide could be detected, indicating the interferences were quite seriously (Fig. 4A). On the contrary, after enrichment with $\text{Fe}_3\text{O}_4@UiO-66-NH_2-Zr^{4+}$ HF, four phosphopeptides were clearly detected with high intensity (Fig. 4B) and the detailed information of the captured phosphopeptides were listed in Table S2 (Supporting information). For human saliva, before enrichment, also no any phosphopeptide signal was detected, and severe signal suppression with high baseline were found (Fig. 4C). Obviously, after enrichment, 21 endogenous

phosphopeptides were detected accompanying the clean baseline signal (Fig. 4D). All peaks were consistent with those reported in the literature [30,38–40]. The enrichment results indicated that the $\text{Fe}_3\text{O}_4@\text{UiO}-66\text{-NH}_2\text{-Zr}^{4+}$ HF could be used as enriched material to capture endogenous phosphopeptides from real biological samples with high efficiency.

A general strategy for thickness-controllable synthesis of MOF-based heterogeneous material with special hollow nanoflowers shapes was demonstrated. The thickness could be freely controlled by changing the assembly cycles. The as-prepared MOF-based heterogeneous material possessed the unique hollow nanoflower structure, well-defined hierarchical pores, good superparamagnetic property, large surface area and good hydrophilicity. Furthermore, after post-modification with Zr^{4+} , the $\text{Fe}_3\text{O}_4@\text{UiO}-66\text{-NH}_2\text{-Zr}^{4+}$ HF showed the ability to selective enrichment of phosphopeptide accompanying with high selectivity, sensitivity and excellent size-exclusion effect. In addition, the $\text{Fe}_3\text{O}_4@\text{UiO}-66\text{-NH}_2\text{-Zr}^{4+}$ HF were successfully applied to specific capture of ultratrace phosphopeptide from complex biosamples. This work will provide a new route for thickness-controllable synthesis of MOF-based heterogeneous material with special hollow nanoflowers shapes, and also shows promising potential in clinical analysis.

Declaration of competing interest

The authors declare no competing interests.

CRediT authorship contribution statement

Ning Zhang: Conceptualization, Writing – original draft. **Mengjie Qin:** Data curation. **Jiawen Zhu:** Investigation, Methodology. **Xuejing Lou:** Formal analysis, Investigation. **Xiao Tian:** Data curation, Methodology. **Wende Ma:** Funding acquisition, Software. **Youmei Wang:** Validation, Visualization. **Minghua Lu:** Funding acquisition, Project administration, Writing – review & editing. **Zongwei Cai:** Writing – review & editing.

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Supplementary materials

Supplementary material associated with this article can be found, in the online version, at doi:10.1016/j.ccl.2024.110177.

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