



## Thiocytochalasins A–D, four sulfur-containing cytochalasins from an endophytic fungus *Phoma multirostrata* XJ-2-1

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### ABSTRACT

Four unprecedented sulfur-containing cytochalasins, thiocytochalasins A–D (**1–4**), were isolated from an endophytic fungus *Phoma multirostrata* XJ-2-1. Thiocytochalasins A (**1**) and B (**2**) feature a novel 5/6/14/5 tetracyclic scaffold, which are the first examples of cytochalasin containing a thiophene moiety. Thiocytochalasins C (**3**) and D (**4**) are epimeric cytochalasin homodimers formed *via* a thioether bridge. Their structures with absolute configurations were established by detailed analysis of the HRESIMS, NMR, and X-ray crystallography. The plausible biogenetic pathway of **1–4** was postulated. Compounds **3** and **4** exhibited significant cytotoxicity against CT26 cells with  $IC_{50}$  values of 0.85 and 0.76  $\mu\text{mol/L}$ , respectively.

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Sulfur-containing natural products are a large class of multifunctional molecules existing in both terrestrial and marine organisms. To date, more than 1000 sulfur-containing natural products have been reported, many of which exhibit potent biological activities and pharmacological properties [1]. For centuries, sulfur has continued to maintain its status as the dominating heteroatom integrated into a set of 362 sulfur-containing FDA approved drugs (besides oxygen or nitrogen) [2]. Sulfonamides, thioethers, sulfones, and penicillin are the most common scaffolds in sulfur containing drugs, which are well studied both on synthesis and application during the past decades [2–4].

Cytochalasins are a group of fungal-derived polyketide-nonribosomal peptide synthetase (PKS-NRPS) natural products characterized by a perhydro-isoindolone core fused with a macrocyclic ring, exhibiting high structural diversity and a broad-spectrum of bioactivities. They are majorly produced by the genera *Aspergillus*, *Chaetomium*, *Xylaria*, *Helminthosporium*, *Penicillium*, *Zygosporium*, *Metarhizium*, *Rosellinia*, *Ascochyta*, *Hypoxylon*, *Spicaria*, and *Phoma* [5–8]. In recent decades, more than 500 cytochalasins have been reported [5,9–15], which are generally classified into six groups, namely cytochalasins, chaetoglobosins, pyrivalasins, aspochalasins, alachalasin, and trichalasin [5,6]. Cytochalasins, a group of cytochalasins with a phenylalanine unit incorporated into the polyketide backbone, is the largest group of cytochalasins [5].

*Phoma* species are a rich source of bioactive secondary metabolites including cytochalasins [16], and the first cytochalasin was isolated from a culture broth of a *Phoma* sp. in 1966 [17]. *Phoma multirostrata* XJ-2-1, an endophytic fungus, was isolated from the medicinal plant *Parasenecio albus* collected from Xinning County, Hunan Province of China. In the previous study, we reported ergo-cytochalasin A, a merocytochalasin consisting of one cytochalasin moiety and one ergosterol moiety with a 5/6/14/6/5/6/6/6 fused octacyclic ring system, and nine new cytochalasins multirostratins B–J, from *P. multirostrata* XJ-2-1 [9,10]. To search for more novel metabolites with unique structures and significant bioactivities, the fungus *P. multirostrata* XJ-2-1 was further investigated, which led to the identification of four unprecedented sulfur-containing cytochalasins, thiocytochalasins A–D (**1–4**) (Fig. 1). To the best of our knowledge, only five sulfur-containing cytochalasins have been reported before [14,18], of which bisaspochalasins B and C are cytochalasin dimers connected by thioether bond [14]. Thiocytochalasins A (**1**) and B (**2**) feature an unprecedented 5/6/14/5 tetracyclic scaffold, which are the first examples of cytochalasin fused with a thiophene moiety. Thiocytochalasins C (**3**) and D (**4**) are epimeric cytochalasin homodimers formed *via* a thioether bridge. Herein, the details of the isolation, structure elucidation, plausible biogenetic pathway, as well as biological activities of thiocytochalasins A–D (**1–4**) are reported.

Thiocytochalasin A (**1**), colorless needles, possessed a molecular formula of  $C_{31}H_{35}NO_5S$  as determined by the HRESIMS ( $[M + Na]^+$ ,  $m/z$  556.2127, calcd. for  $C_{31}H_{35}NO_5SNa$ , 556.2134), requiring 15 double bond equivalents (DBEs), which was consistent with the  $^{13}C$

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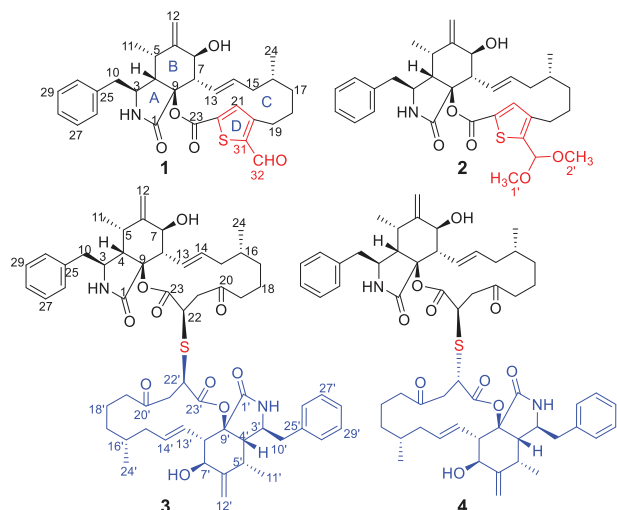


Fig. 1. Chemical structures of compounds 1–4.

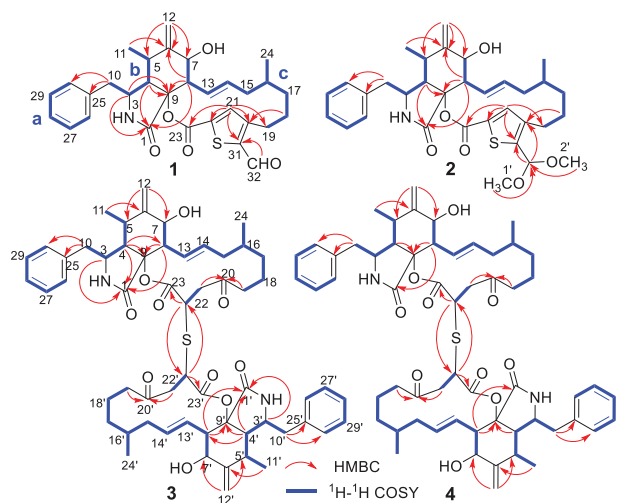


Fig. 2. Key HMBC and  $^1\text{H}$ – $^1\text{H}$  COSY correlations of 1–4.

NMR data. Its IR spectrum showed absorptions for hydroxy ( $3419\text{ cm}^{-1}$ ), ester carbonyl ( $1717\text{ cm}^{-1}$ ), conjugated aldehyde and/or amide ( $1668\text{ cm}^{-1}$ ), and phenyl ( $1604$  and  $1494\text{ cm}^{-1}$ ) groups. The  $^1\text{H}$  NMR data (Table S1 in Supporting information) of **1** showed two doublet methyls at  $\delta_{\text{H}}$  1.01 (d, 3H,  $J=6.7\text{ Hz}$ ) and 1.15 (d, 3H,  $J=6.6\text{ Hz}$ ); five olefinic protons at  $\delta_{\text{H}}$  5.23 (br s, 1H), 5.47 (br s, 1H), 5.77 (ddd, 1H,  $J=15.2, 10.6, 3.4\text{ Hz}$ ), 5.90 (ddd, 1H,  $J=15.2, 9.6, 2.2\text{ Hz}$ ), and 7.93 (s, 1H); one monosubstituted phenyl group with five coupled aromatic protons at  $\delta_{\text{H}}$  7.24 (d, 2H,  $J=7.6\text{ Hz}$ ), 7.25 (m, 1H), and 7.33 (t, 2H,  $J=7.6\text{ Hz}$ ); one aldehyde proton at  $\delta_{\text{H}}$  10.05 (s, 1H). The  $^{13}\text{C}$  NMR data (Table S1), with the aid of DEPT and HSQC experiments, exhibited 31 carbon resonances, comprising three carbonyls ( $\delta_{\text{C}}$  160.8, 170.1, and 182.7), six aromatic carbons, eight olefinic carbons, one  $\text{sp}^3$  quaternary carbons, six  $\text{sp}^3$  methines, five  $\text{sp}^3$  methylenes, and two methyls. These functionalities accounted for 11 out of the 15 DBEs, indicating four additional rings in the structure of **1**. The aforementioned NMR data, along with the previous reports of the chemical investigation on *P. multirostrata* EA-12 [9,10], suggested that **1** was likely a cytochalasin.

The planar structure of **1** was established by interpretation of its 2D NMR spectra (Fig. 2). Analysis of the  $^1\text{H}$ – $^1\text{H}$  COSY spectrum of **1** revealed the presence of three spin coupling fragments **a**–**c** which are shown with blue bold bonds (Fig. 2). Then, those fragments with the nonprotonated carbons and heteroatoms were

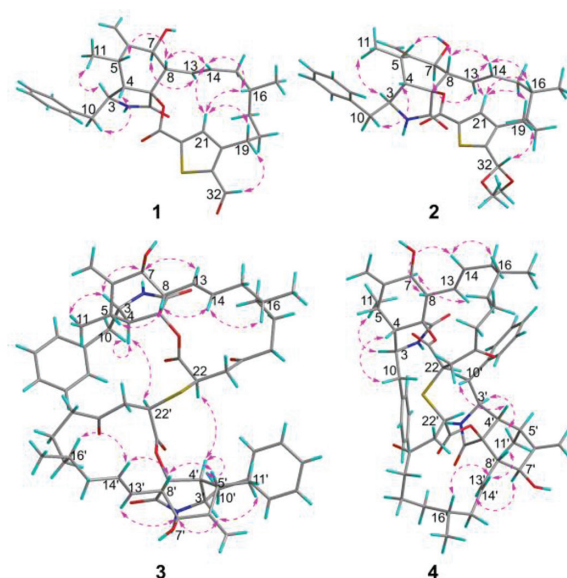


Fig. 3. Key NOESY correlations of 1–4.

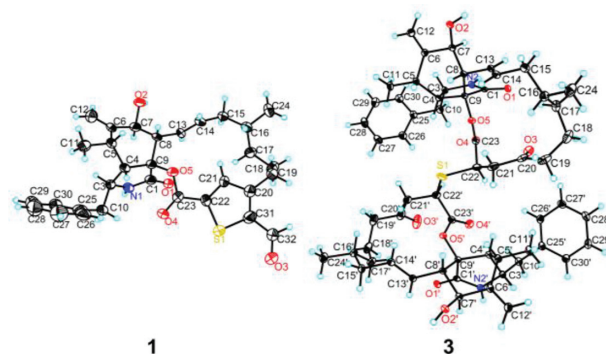


Fig. 4. X-ray ORTEP drawing of compounds **1** and **3**. Displacement ellipsoids are drawn at the 50% probability level.

assembled by examination of the HMBC spectrum to furnish the planar structure of **1**. The HMBC cross-peaks of  $\text{H}_2$ –10/C-25 and C-30 indicated the monosubstituted phenyl group attached to C-10. The HMBC correlations of H-3/C-1, C-9, H-4/C-1, C-9, H<sub>3</sub>-11/C-6, H<sub>2</sub>–12/C-5, C-6, C-7, H-8/C-1, C-6, C-9 delineated the perhydroisoindolone core (rings A and B). The construction of ring C was established by the HMBC networks of H<sub>2</sub>–18/C-20, H-21/C-19, C-22, C-23. The HMBC cross-peaks of H-21/C-20, C-22, C-31, H-32/C-31, allowed the establishment of the unique thiophene ring D, which bridged to the ring C at C-20 and C-22, and simultaneously suggested that the aldehyde group is attached to C-31. Thus, the gross structure of **1** was elucidated as shown in Fig. 1.

The relative configuration of compound **1** was fixed by the analysis of the  $^1\text{H}$ – $^1\text{H}$  coupling constants (Table S1) and the NOESY correlations (Fig. 3). The *E*-geometry of  $\Delta^{13}$  double bond was deduced from the large coupling constant between H-13 and H-14 ( $J_{13,14}=15.2\text{ Hz}$ ). The NOESY correlations of H-3 with H<sub>3</sub>–11 revealed that H-3 and H<sub>3</sub>–11 were positioned on the same side of the molecule and randomly assigned as  $\alpha$ -oriented, while H-4, H-5, H-8 and H-16 were assigned to be  $\beta$ -oriented by the NOESY cross-peaks of H-4/H<sub>2</sub>–10; H-8/H-5 and H14; H-14/H-16. The  $\alpha$  orientation of H-7 was determined by the NOESY correlation between H-7 and H-13. To confirm the structure assigned for compound **1** and determine its absolute configuration, quality crystals were successfully obtained from recrystallization in  $\text{CH}_2\text{Cl}_2/\text{MeOH}/\text{H}_2\text{O}$

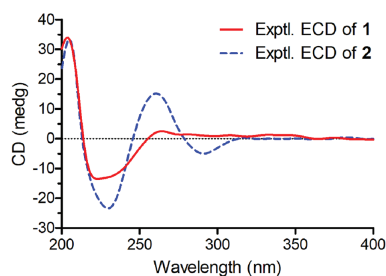


Fig. 5. Experimental ECD spectra of **1** and **2**.

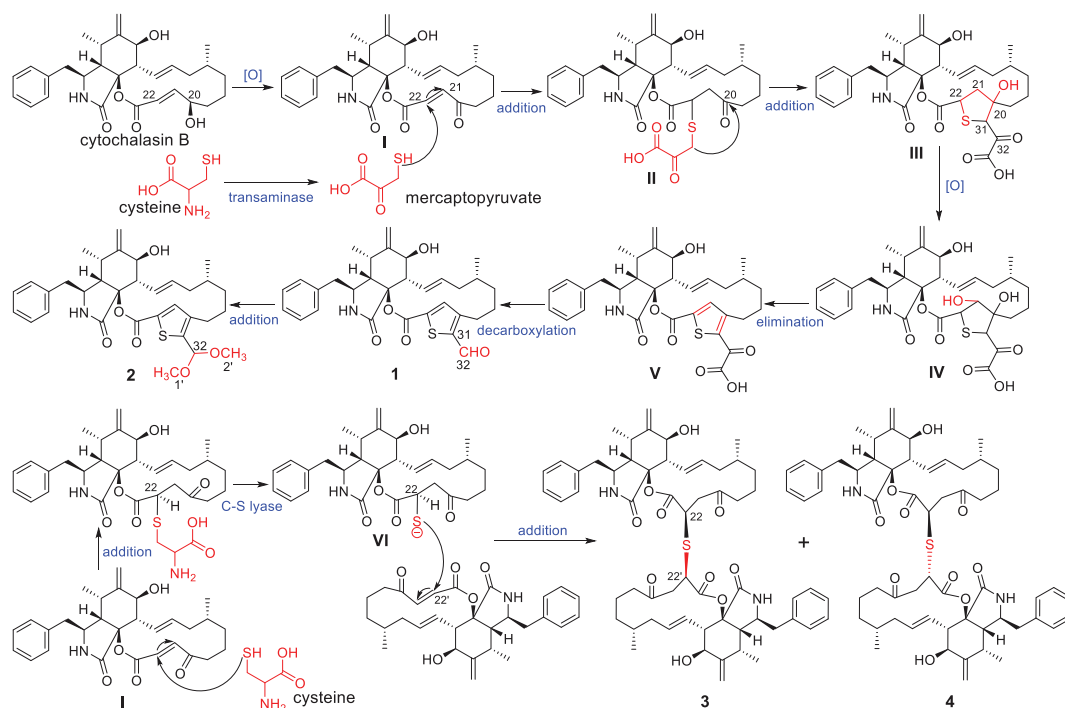
(10:10:1) at room temperature, and then subjected to single-crystal X-ray diffraction (XRD) analysis using Ga  $K\alpha$  radiation (Fig. 4). The X-ray crystallography result not only secured the above structural assignment for **1** but also established its absolute configuration (3*S*, 4*S*, 5*S*, 7*S*, 8*S*, 9*S*, 16*R*) by the excellent Flack parameter [0.12 (2)].

Thiocytochalasin B (**2**) was obtained as white powder. Its molecular formula was deduced to be  $C_{33}H_{41}NO_6S$  from (+)-HRESIMS ion peak at  $m/z$  602.2542  $[M+Na]^+$  (calcd. for  $C_{33}H_{41}NO_6SNa$ , 602.2552) and the  $^{13}C$  NMR data (Table S1), indicating 14 DBEs. The IR spectrum revealed the presence of hydroxy ( $3383\text{ cm}^{-1}$ ), carbonyl ( $1711\text{ cm}^{-1}$ ), and terminal double bond ( $1640$  and  $972\text{ cm}^{-1}$ ). Detailed comparison of the NMR data between **1** and **2** (Table S1) indicated they had identical carbon skeleton. The main difference was that the C-32 aldehyde group in **1** was replaced by a dimethoxy methine group in **2**, which was confirmed by the HMBC correlations (Fig. 2) from H<sub>3</sub>-1' ( $\delta_H$  3.36) and H<sub>3</sub>-2' ( $\delta_H$  3.32) to C-32 ( $\delta_C$  98.7). The NOESY correlations of **2** disclosed that this compound had the same relative configuration as that of compound **1** (Fig. 3). Finally, the absolute configuration of **2** was identified by comparison of its ECD spectrum with that of **1** (Fig. 5).

Thiocytochalasin C (**3**) was isolated as colorless needles with the molecular formula of  $C_{58}H_{72}N_2O_{10}S$  from its HRESIMS ion peak at  $m/z$  1011.4811  $[M+Na]^+$  (calcd. for  $C_{58}H_{72}N_2O_{10}SNa$ , 1011.4805),

requiring 24 DBEs. The  $^{13}C$  NMR (Table S2 in Supporting information) displayed only 29 resonances for **3**, suggesting that compound **3** was a symmetrical cytochalasin dimer linked by a central bridged sulfur atom. Its IR absorption bands at  $3359$  and  $1717\text{ cm}^{-1}$  suggested the presence of hydroxy and carbonyl groups, respectively. Further analysis of the 1D NMR data (Table S2) revealed that the cytochalasin monomer in **3** closely resembled the known compound cytochalasin B<sub>6</sub> [19], with the only difference being the replacement of a methylene ( $\delta_C$  29.0, C-22) in cytochalasin B<sub>6</sub> by a methine ( $\delta_C$  42.1, C-22) in **3**. This was confirmed by the  $^1H$ - $^1H$  COSY correlation of H<sub>2</sub>-21/H-22, and the HMBC cross-peaks from H<sub>2</sub>-19 and H<sub>2</sub>-21 to C-20, and from H-22 to C-23 (Fig. 2). Considering the symmetry of the molecule, this change led to the deduction that the two monomers were connected by virtue of a sulfur bridge between C-22 and C-22', which was supported by the HMBC correlation of H-22/C-22' (Fig. 2). The relative configuration of **3** was determined by analysis of the NOESY spectrum (Fig. 3). The NOESY correlations of H-3/H-7 and H<sub>3</sub>-11 indicated the assignment of the  $\alpha$ -configuration for these protons. The  $\beta$ -configuration for H-4, H-8 and H<sub>3</sub>-16 was fixed according to the NOESY correlations of H-4/H-10; H-8/H-5, H-14; and H-14/H-16. However, the relative configuration of C-22 could not be resolved because of the absence of specific evidence. Fortunately, a suitable crystal was obtained from a solution of  $CH_2Cl_2/MeOH$  (1:1). The single-crystal X-ray diffraction analysis using Cu  $K\alpha$  radiation (Fig. 4) with Flack parameter [0.009 (10)] corroborated the structural assignment of **3** based on NMR and identified its absolute configuration as 3(3')*S*, 4(4')*S*, 5(5')*S*, 7(7')*S*, 9(9')*S*, 16(16')*R*, 22(22')*R*.

Thiocytochalasin D (**4**), white powder, had the same molecular formula of  $C_{58}H_{72}N_2O_{10}S$  as **3**, as deduced from its HRESIMS ion peak at  $m/z$  1011.4821  $[M+Na]^+$  (calcd. for  $C_{58}H_{72}N_2O_{10}SNa$ , 1011.4805) and the  $^{13}C$  NMR data (Table S3 in Supporting information). Its IR absorption bands at  $3421$  and  $1713\text{ cm}^{-1}$  suggested the presence of hydroxy and carbonyl groups, respectively. A detailed comparison of the NMR data of **4** and **3** (Tables S2 and S3) indicated that the two compounds are structurally related, except that the structural symmetry present in **3** was broken in **4**, as the  $^{13}C$



Scheme 1. Plausible biosynthetic pathway of compounds **1**–**4**.

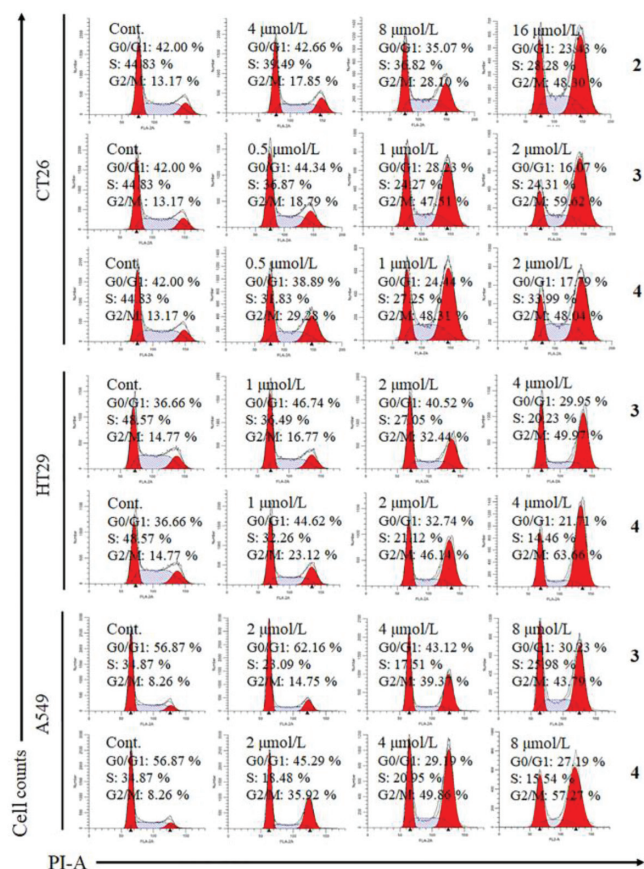


Fig. 6. Cell cycle analysis of CT26, HT29 and A549 cells after 24 h treatment with compounds 2–4 at different concentrations using PBS as a negative control (Cont.).

NMR of **4** showed 58 signals. Further analysis of the 2D NMR spectra (Fig. 2) constructed the planar structure of **4**, which was the same as that of **3**. The only difference between **4** and **3** was the opposite configuration at C-22', which was determined by comparison of their chemical shifts and coupling constants as H-22' at  $\delta_{\text{H}}$  3.91 (dd, 2H,  $J=11.4, 3.1$  Hz) for **3** while H-22' at  $\delta_{\text{H}}$  4.14 (dd, 1H,  $J=7.8, 3.5$  Hz) for **4**. The relative configuration of other chiral centers in **4** was assigned as the same as that of **3** by their similar NMR data and NOESY correlations (Fig. 3). As cytochalasins shared the same biogenetic pathway, the absolute configuration of **4** was assigned as 3(3')S, 4(4')S, 5(5')S, 7(7')S, 9(9')S, 16(16')R, 22R, 22'S. The structure of **4** was hence established as shown in Fig. 1.

The plausible biosynthesis pathway of thiocytochalasins A–D (**1–4**) is proposed in Scheme 1. On the basis of the systematic report of enzymatic carbon-sulfur bond formation in natural product biosynthesis [20], cysteine is considered to be the sulfur source of thiocytochalasins A–D (**1–4**). And cytochalasin B, a main cytochalasin from this strain [9], is postulated to be the precursor. First, cytochalasin B was reduced to give a vital intermediate **I** via an oxidation reaction, which subsequently involved a nucleophilic addition to C-22 in the  $\alpha,\beta$ -unsaturated 1,4-diketone moiety of intermediate **I** with mercaptopyruvate to produce the thio intermediate **II**. The intermediate **II** underwent a series of reactions including addition, oxidation, elimination, and decarboxylation to generate thiocytochalasin A (**1**) with a 2,3,5-trisubstituted thiophene moiety. Then, thiocytochalasin B (**2**) was produced by an addition of **1**. The cysteine could conduct a nucleophilic addition to intermediate **I**, leading to an intermediate **VI** which could generate a sulfur anion to couple with another monomer intermediate **I** to produce a pair

of epimeric cytochalasin homodimers thiocytochalasins C (**3**) and D (**4**) decorated with a thioether bridge.

Compounds **1–4** were evaluated for their cytotoxicity against MCF-7 human breast adenocarcinoma, HepG2 human hepatocellular carcinoma, CT26 murine colon carcinoma, HT-29 human colon carcinoma, and A549 human lung carcinoma cell lines using MTT method. As a result, compounds **1** and **2** showed modest cytotoxicity against the tested cell lines with  $\text{IC}_{50}$  values of 4.17–23.41  $\mu\text{mol/L}$  (Table S4 in Supporting information). Compounds **3** and **4** displayed strong cytotoxicity against four cell lines with  $\text{IC}_{50}$  values from 0.76  $\mu\text{mol/L}$  to 7.52  $\mu\text{mol/L}$ , and exhibited moderate cytotoxicity against HepG2 cells (Table S4). Subsequently, the effects of **2–4** on cell cycle progression of CT26, HT29 and A549 cells were investigated. The results showed that they could concentration-dependently induce G2/M cell cycle arrest in these cells, especially compounds **3** and **4** could significantly arrest cell cycle G2/M phase of CT26 cells at the concentration of 1  $\mu\text{mol/L}$  (Fig. 6).

In conclusion, the discovery of thiocytochalasins A–D (**1–4**) from an endophytic fungus *P. multirostrata* XJ-2-1 has expanded the scaffolds of cytochalasins with sulfur-containing moiety. In particular, identification of the thiocytochalasins A (**1**) and B (**2**), first examples of cytochalasins fused with a thiophene moiety, is of great importance in the research field of cytochalasins. It is noteworthy that two cytochalasin homodimers with thioether bridge have been reported from *Aspergillus flavipes* before, and thiocytochalasins C (**3**) and D (**4**) exhibit their differences on the type of cytochalasin monomer and non-selective addition at C-22' in the  $\alpha,\beta$ -unsaturated 1,4-diketone moiety compared to the reported compounds bisaspochalasins B and C [14]. In addition, the significant cytotoxicity of compounds **3** and **4** against five cancer cell lines especially against CT26 cells with  $\text{IC}_{50}$  values of 0.85 and 0.76  $\mu\text{mol/L}$ , respectively, is better than the previously reported cytochalasin monomer from this strain [9], which might provide an idea to boost the discovery of antitumor candidates for pharmaceutical chemists.

#### Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper

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#### Supplementary materials

Supplementary material associated with this article can be found, in the online version, at doi:10.1016/j.ccllet.2022.03.063.

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