



A tandem asymmetric oxidation-oxa-Michael sequence for dearomatization of β -naphthols



Linqing Wang, Haiyong Zhu, Tianyu Peng, Yingfan Xu, Yanzhe Hou, Shixin Li, Shiming Pang, Hailong Zhang*, Dongxu Yang*

Key Laboratory of Preclinical Study for New Drugs of Gansu Province, Institute of Drug Design & Synthesis, School of Basic Medical Sciences, Lanzhou University, Lanzhou 730000, China

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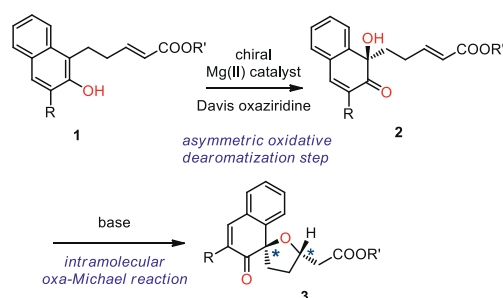
ABSTRACT

A catalytic asymmetric hydroxylative dearomatization reaction has been disclosed, and the products can smoothly transform into spiroannulation adducts by simply treated with a base under mild conditions. Novel *in-situ* generated magnesium catalytic methods are developed by application of combinational ligands. Related concise transformaitons of the spiroannulation adducts have been carried out.

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Catalytic asymmetric dearomatization (CADA) reaction is a highly attractive protocol in recent years for chemists, because it is a straightforward strategy to effectively build three-dimensional chemical structures from aromatic compounds [1–11]. In this context, the dearomatizative spiroannulation reactions of phenols are also rapidly developed for its powerful ability on construction of spirocyclic molecules. There are several representative methods for dearomatizative spiroannulation reactions, for example, one is the intramolecular pathway, including oxidation and substitution reactions [12–21]; another pathway can be intermolecular cyclization reactions, such as the catalytic asymmetric [1 + 4] and [2 + 3] spiroannulations reported very recently [22–25].

The catalytic asymmetric oxidative dearomatization reaction of phenols can break the aromaticity to establish the stereochemical structure by simply equipping a hydroxyl group to the molecules. Several catalytic methods, for example, hypervalent iodine(iii) reagent [12–17], diamine-copper [26] or *N,N'*-dioxide-scandium catalysts [27] have been successfully developed for the asymmetric hydroxylative dearomatization of phenol compounds. However, the generated chiral tertiary alcohols have less been utilized for further intramolecular additions to construct more complicated skeletons [28], such as oxaspirocyclic molecules. Herein, on the basis of recent works on asymmetric dearomatization reac-



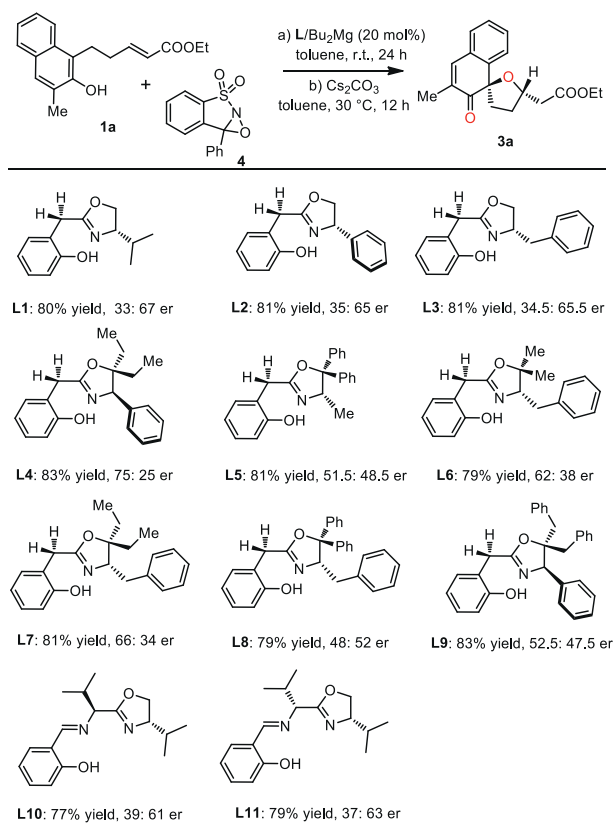
Scheme 1. The designed tandem asymmetric oxidation-oxa-Michael reaction.

tions of phenol compounds [29–32], we designed a tandem asymmetric oxidation-oxa-Michael reaction sequence for dearomatization of β -naphthols **1**, by developing *in-situ* generated magnesium catalytic methods with combinational use of different ligands [33–51]. The current reaction including a Mg(II)-catalyzed enantioselective oxidation step to generate the hydroxylation adduct **2** and a base promoted intramolecular oxa-addition step to finish the formal spiroannulation process (Scheme 1) [52–56].

In the initial experiment of the tandem asymmetric oxidation-oxa-Michael reaction between naphthol derivative **1a** and Davis oxaziridine **4** [57,58], varies of *in-situ* generated magnesium catalysts employing of different chiral ligands were examined, and Cs_2CO_3 was identified as the optimal base for the stepwise in-

* Corresponding authors.

E-mail addresses: zhanghl@lzu.edu.cn (H. Zhang), yangdx@lzu.edu.cn (D. Yang).



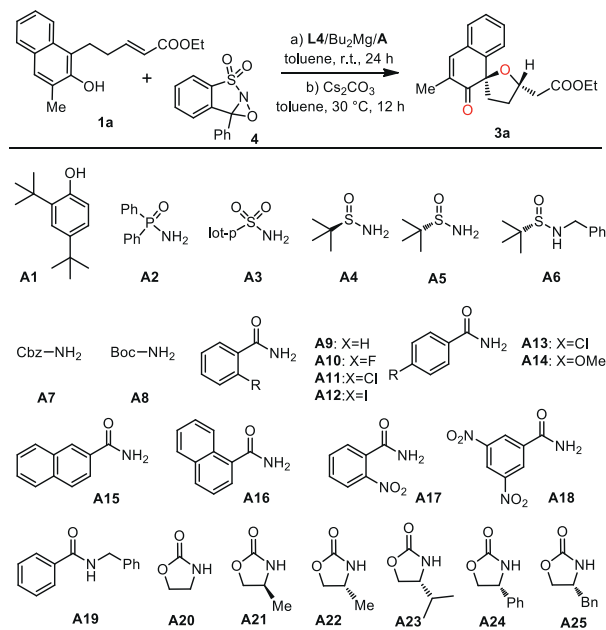
Scheme 2. Initial optimization process of the tandem asymmetric oxidation-oxa-Michael reaction. Reactions were performed with naphthol **1a** (0.10 mmol), **4** (0.10 mmol) in toluene (1.0 mL) in the presence of Bu_2Mg (20 mol%) and **L4** (20 mol%).

tramolecular oxa-Michael reaction, resulting in good yields and diastereoselectivities (>20:1 dr). The employment of BINOLs or Salen as chiral ligands revealed that these magnesium catalysts could mediate the dearomatizative hydroxylation reaction of naphthol **1a**, and afforded the oxidative dearomatization product **3a** in excellent yields but with disappointed enantioselectivities. The detail optimization process is deposited in Supporting information. Further evaluation of oxazoline-OH ligands with different substituents and backbones indicated that all the reactions proceeded smoothly to form the desired dearomatization product **3a** with variable enantioselective control (Scheme 2). We found the ligands with different chiral substituents gave similar results (Scheme 2, **L1–L3**), some poly-substituted ligands were also introduced and led to dramatically different enantioselectivities. The screening progress finally identified **L4** as a more promising chiral ligand, which was derived from a multi-substituted amino alcohol, leading to the oxidative dearomatization product **3a** in an excellent yield with a moderate enantioselectivity (83% yield, 75:25 er).

To further improve the enantioselectivities of this oxidative dearomatization reaction, we turned to introduce a second ligand and tested its effects on the magnesium catalysis. As illustrated in Table 1, a wide scope of additives were examined as the second ligands in the Mg(II) mediated dearomatization process. The introduction of phenol such as **A1** obviously decreased er value (Table 1, entry 1). The introduction of phosphamide or sulfamide did not show promise results (Table 1, entries 2 and 3). Utilization of various of chiral sulfonamide obviously improved the er values, and we found some benzamides could also increase the enantioselectivities in the presence of **L4**/ Bu_2Mg , and the use of 1-naphthamide **A16** gave a relatively better enantioselectivity but with slightly lower yield. To our delight, when oxazolidone **A20** was employed as the

Table 1

The optimization process of combinational magnesium catalyst for the reaction.^a



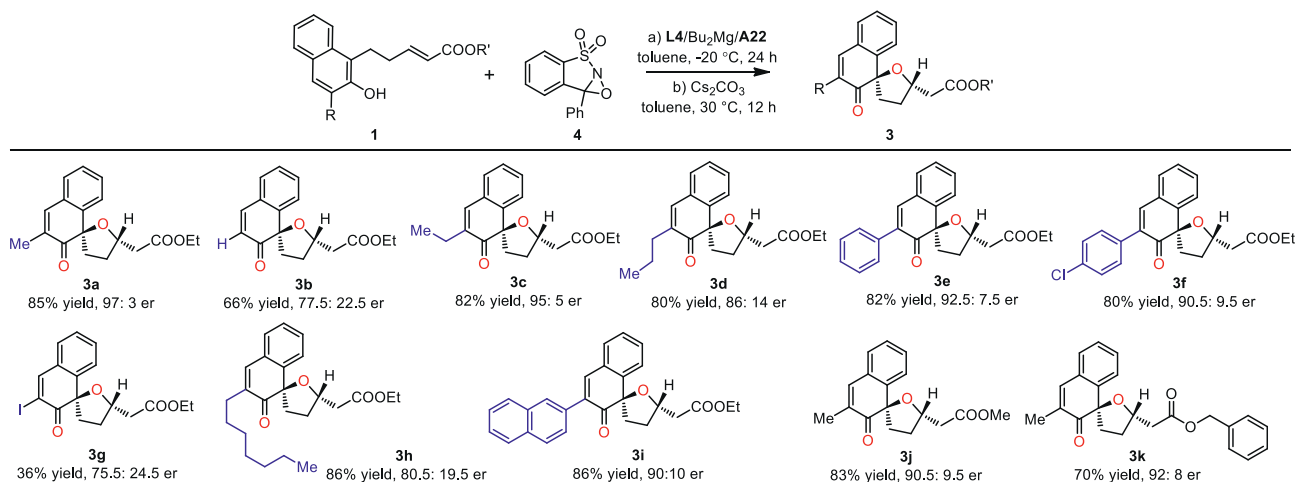
Entry	A	Yield (%)	er
1	A1	60	66.5:33.5
2	A2	86	60:40
3	A3	59	69:31
4	A4	71	86:14
5	A5	70	83.5:16.5
6	A6	65	77.5:22.5
7	A7	80	80:20
8	A8	84	78:22
9	A9	85	83.5:16.5
10	A10	78	72.5:27.5
11	A11	80	78:22
12	A12	78	77.5:22.5
13	A13	75	81:19
14	A14	80	84:16
15	A15	75	78:22
16	A16	67	87.5:12.5
17	A17	54	83.5:16.5
18	A18	70	74.5:25.5
19	A19	73	74.5:25.5
20	A20	87	83.5:16.5
21	A21	85	83:17
22	A22	87	89:11
23	A23	79	79.5:20.5
24	A24	86	79.5:20.5
25	A25	82	79:21
26 ^b	A22	85	92:8
27 ^c	A22	85	97:3

^a Reactions were performed with naphthol **1a** (0.10 mmol), **4** (0.10 mmol) in toluene (1.0 mL) in the presence of Bu_2Mg (20 mol%) and **L4** (20 mol%) and the second ligand **A** (20 mol%).

^b Reaction was carried out with **L4** (30 mol%) and Bu_2Mg (20 mol%).

^c Reaction was carried out with **L4** (30 mol%), Bu_2Mg (20 mol%) and **A22** (40 mol%) in toluene (2.0 mL).

additive ligand, the er value of the model reaction increased obviously to 83.5:16.5 (Table 1, entry 20). Encouraged by this result, a series of optically substituted oxazolidones were further examined and the ligand **A22** led to relatively better results. Interestingly, it was found that the addition of (*R*)-oxazolidone **A22** resulted in slight higher er value than its optical isomer **A21**, and the increase of steric hindrance of the substituents on (*R*)-oxazolidones would lead to decreased er values of the oxidative dearomatization product **3a** (Table 1, entries 23–25). Further increasing the loading



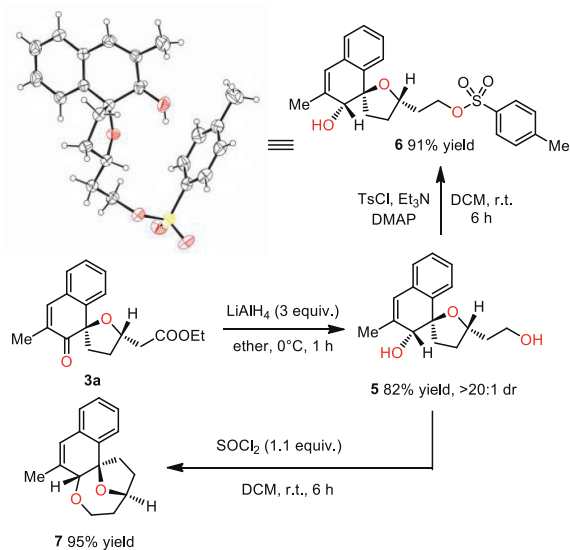
Scheme 3. Substrate scope of the reaction. Reactions were performed with naphthol **1a** (0.20 mmol), **4** (0.20 mmol) in toluene (2.0 mL) in the presence of Bu₂Mg (20 mol%) and **L4** (30 mol%) and the second ligand **A22** (40 mol%).

amount of **L4** and **A22** led to the optimized results of the tandem asymmetric oxidation-oxa-Michael reaction.

With the development of suitable conditions and identification of combinational magnesium catalyst generating from **L4**, **A22** and Bu₂Mg, we next began to investigate the substrate scope of this tandem asymmetric oxidation-oxa-Michael reaction. As the results summarized in Scheme 3, different alkyl substituted (Me, Et, *n*-Pr and *n*-heptyl) β -naphthols were well tolerated, and their corresponding dearomative products were obtained in excellent yields and good enantioselectivities (**3a**, **3c**, **3d** and **3h**). β -Naphthols bearing aryl substituents at C3 position, challenging substrates in some intermolecular asymmetric oxidative dearomatization reactions, were also worked well and led to the desired spirocyclic hydrofuran structures in good yields and enantioselectivities (**3e**, **3f**, **3i**). Moreover, naphthols with halogen or hydrogen atom at the C3 position, which were often failure substrates in dearomatization reactions of phenol compounds, were also tolerable but resulted in relatively lower enantioselectivities (**3b**, **3g**). Furthermore, two other substrates with different unsaturated esters equipped at the C1 position also underwent the dearomative reactions smoothly with satisfied enantioselectivities (**3j**, **3k**).

Next, several transformations of the spiroannulation adducts were carried out (Scheme 4). The ester group and carbonyl group in the cyclohexanone frameworks of **3a** could be readily reduced by lithium aluminum hydride to the diol product **5** in excellent yield and diastereoselectivities, and the terminal alcohol was selectively protected with *p*-toluene sulfonyl group by simply treated with tosyl chloride and triethylamine in the presence of DMAP, led to the compound **6**, and the absolute structure was analyzed by corresponding X-ray crystallographic studies. Moreover, compound **7** with an oxygen bridged ring was obtained in a high yield by simply treating **5** with SOCl₂ at mild conditions.

On the basis of our previous works on developing of *in-situ* generated magnesium catalysts with the utilization of oxazoline-OH chiral ligands [29,31,46,48], a hypothesized mechanism of this Mg(II)-catalyzed hydroxylative reaction is proposed (Scheme 5). The magnesium compound **I** was confirmed in our former works and we speculated the bidentate coordinate phenol **1** is prone to finish another neutralization step to generate the complex **II**. The oxaziridine might coordinate to the magnesium centre in the less steric hindered direction and the oxazolidone **A22** can occupy another possible coordination site, which might be the reason to improve the enantioselectivities (**III**). Then the hydroxylative step occur at the favored *Si*-face to generate the dearomatization product.



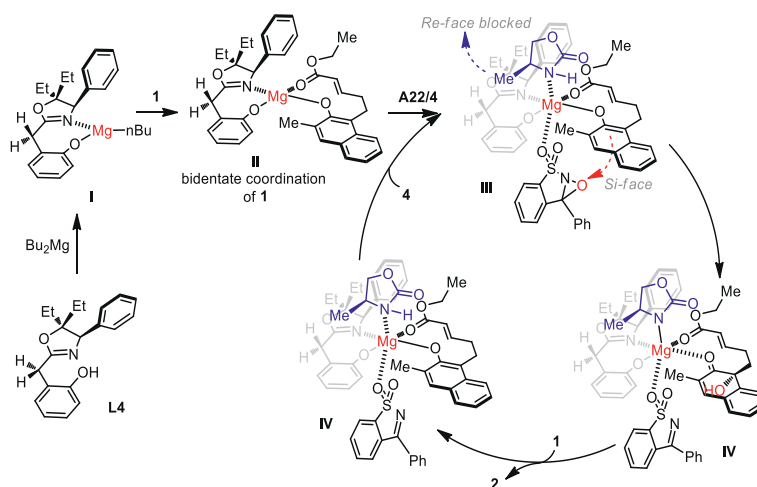
Scheme 4. Transformation of the formal spiroannulation product.

Next release the adduct **2** and the generated imine initiate another catalytic cycle after replacement with the bidentate phenol **1** and another molecule of oxaziridine.

In summary, we have developed a combinational magnesium catalytic method for the tandem asymmetric dearomatization oxidation-oxa-Michael reaction of the naphthol compounds. The introduction of additives such as oxazolidones can obviously improve the enantioselectivities. Representative substituents on the naphthols were examined and led to satisfied results in this reaction. And several brief transformations of the spiroannulation adducts were carried out in good results. Our future work will focus on developing of *in-situ* generated magnesium catalysts and applying them in more catalytic asymmetric reactions.

Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.



Scheme 5. A hypothesized mechanism of the asymmetric hydroxylative dearomatization reaction.

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Supplementary materials

Supplementary material associated with this article can be found, in the online version, at doi:10.1016/j.ccllet.2021.12.075.

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