



## Communication

# One-pot tandem route to fused indolizidines and quinolizidines: Application in the synthesis of alkaloids and bioactive compounds

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## ABSTRACT

Fused indolizidines and quinolizidines are important skeletons in a variety of natural products and pharmacologically important compounds. A one-pot tandem route from amide to fused indolizidines and quinolizidines is disclosed. This method is conducted in mild conditions and shows well tolerance of functional groups. It is also easy to be scaled up to gram scale and can be applied smoothly to the total synthesis of alkaloids such as (±)-crispine A, (±)-xylopinine, (±)-desbromoarborescidine A, (±)-harmicine and other bioactive substances.

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Fused indolizidines and quinolizidines are found in a variety of natural products as well as in pharmacologically important compounds and possess a wide spectrum of biological activities [1–5] (Fig. 1).

Inspired by the important value of these compounds in modern pharmacology [6,7], scholars have devoted significant efforts to their construction [8–15]. Among them, Bischler–Napieralski reaction (B–N reaction) [16,17] offers a very important strategy (Scheme 1a). However, harsh reaction conditions or highly corrosive reagents such as POCl<sub>3</sub> and P<sub>2</sub>O<sub>5</sub> are involved in traditional B–N reaction. Moreover, to obtain the target fused ring, multiple steps of reactions are imperative, leading to reduced yields and tedious operation due to the indispensable isolation and purification of intermediates [8,9]. Although some other existing methods are efficient, they also suffer from some inevitable drawbacks, such as using hazardous reagents [10], expensive metal catalysts [11,12] or commercially unavailable catalysts or substrates [13,9–15].

In recent decades, amide activation has become an emerging tool for chemoselective synthesis [18,19]. Researchers such as

Movassaghi [20,21] and Huang [22] have conducted elegant works in this field.

In 2008, Movassaghi's group reported a method to synthesize isoquinoline and β-carboline derivatives through a mild B–N reaction by using trifluoromethanesulfonic anhydride (Tf<sub>2</sub>O) to activate amide [20]. But unsubstituted indole derivatives were not suitable for this method to synthesize corresponding β-carboline derivatives, which limited its application scope to a certain extent. And the synthesis of structurally more complicated fused indolizidines and quinolizidines was also not involved (Scheme 1b).

As part of our program to develop practical method for the construction of pharmaceutical products [23,24], herein, a one-pot tandem route [25–27] based on amide activation to fused indolizidines and quinolizidines is disclosed (Scheme 1c).

5-Chloro-*N*-(4-methoxyphenethyl)pentanamide (**6a**) was selected as the starting material for optimization of reaction conditions. The intermediate dihydroisoquinoline other than the desired tricyclic product (**7a**) was detected under the initial reaction conditions (Table 1, entry 1). This finding suggested that the activation of amide and B–N reaction went well, but reduction following cyclization did not occur. This phenomenon might be due to the poor solubility of the reductant NaBH<sub>4</sub> in the aprotic solvent CH<sub>2</sub>Cl<sub>2</sub>. The addition of methanol solved the problem satisfactorily, giving the target product in 69% yield (Table 1, entry 2). We then screened potential bases (Table 1, entries 3–7), and

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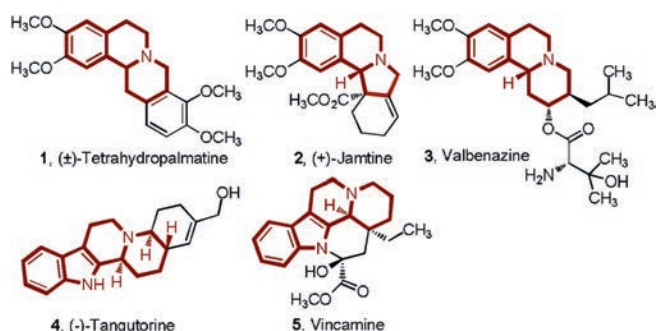
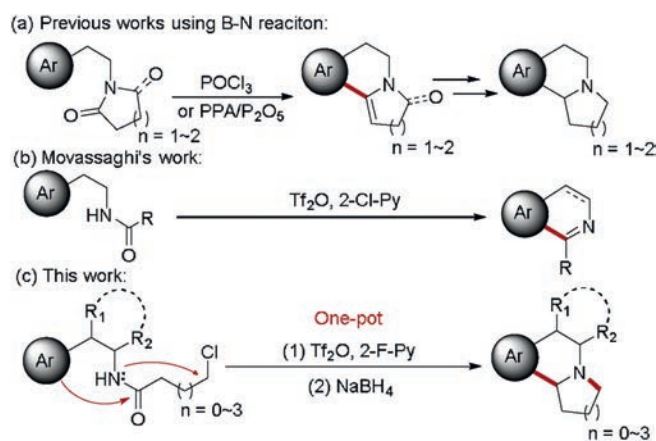


Fig. 1. Related natural products and drugs containing fused indolizidine or quinolizidine structures.



Scheme 1. Comparison of previous works and this work.

Table 1  
Optimization of reaction conditions.<sup>a</sup>

Entry	Base (equiv.)	Activation T (°C)	Reductant	Yield (%) <sup>b</sup>
1	2-Cl-Py (1.2)	-30	NaBH <sub>4</sub>	ND <sup>c</sup>
2	2-Cl-Py (1.2)	-30	NaBH <sub>4</sub>	69
3	-	-30	NaBH <sub>4</sub>	12
4	2-F-Py (1.2)	-30	NaBH <sub>4</sub>	75
5	2-I-Py (1.2)	-30	NaBH <sub>4</sub>	61
6	Pyridine (1.2)	-30	NaBH <sub>4</sub>	55
7	3,5-dimethylpyridine (1.2)	-30	NaBH <sub>4</sub>	58
8	2-F-Py (2.0)	-30	NaBH <sub>4</sub>	73
9	2-F-Py (1.2)	0	NaBH <sub>4</sub>	66
10	2-F-Py (1.2)	-78	NaBH <sub>4</sub>	93 (90)
11	2-F-Py (1.2)	-78	KBH <sub>4</sub>	86
12	2-F-Py (1.2)	-78	NaBH <sub>3</sub> CN	80
13	2-F-Py (1.2)	-78	NaBH(OAc) <sub>3</sub>	76

<sup>a</sup> Reaction conditions: **6a** (0.5 mmol), Tf<sub>2</sub>O (1.1 equiv.), base, CH<sub>2</sub>Cl<sub>2</sub> (5 mL), Ar, activation temp., 30 min → 40 °C, 2 h → reductant (2.0 equiv.), MeOH (5 mL), r.t., 2 h.

<sup>b</sup> Yields were determined by <sup>1</sup>H NMR with 1,1,2,2-tetrachloroethane as the internal standard. Yield in parentheses is the isolated yield.

<sup>c</sup> No reaction. Without MeOH on the reduction stage.

2-fluoropyridine (2-F-Py) was found to be the most efficient (Table 1, entry 4). Nevertheless, increasing the amount of 2-F-Py did not contribute to the improvement 9 and 10 of yield (Table 1, entry 8). When base was absent, the reaction also occurred in spite of reduced yield (Table 1, entry 3). The activation temperature was subsequently investigated (Table 1, entries 9–10). With the activation temperature decreased to -78 °C, the product was obtained in 93% yield (Table 1, entry 10). Other different reductants including KBH<sub>4</sub>, NaBH<sub>3</sub>CN, and NaBH(OAc)<sub>3</sub> were then tested, but the outcomes were inferior to NaBH<sub>4</sub>.

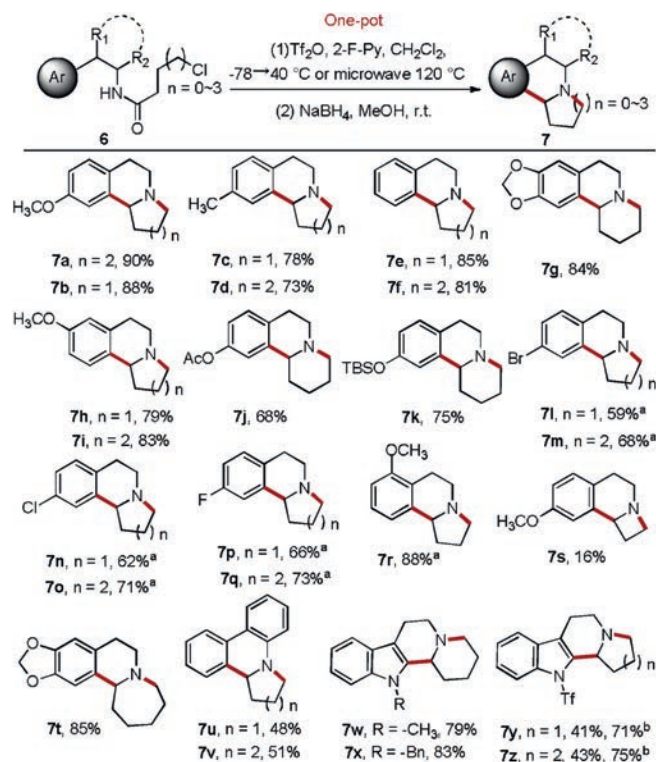
We examined the substrate scope under the optimized conditions. Substrates with electron-donating substitutions or without any substitution in the aromatic nucleus all reacted smoothly to generate corresponding fused indolizidines and quinolizidines in good to excellent yields (Scheme 2, **7a–7k**). The reaction system was applied to substrates containing somewhat electron-withdrawing groups. However, no desired tricyclic products were detected. This was probably because the reduced electron density on the aromatic ring led to difficulty in the occurrence of the B–N reaction. Nevertheless, with the aid of microwave, tricyclic compounds with somewhat electron-withdrawing groups such as F-, Cl- and Br- were obtained in moderate to good yields (Scheme 2, **7l–7q**). The substitution position also affected the reaction. The reaction with the substitute group in *para* and *meta* positions proceeded smoothly under the regular condition, whereas that with *ortho* substitution reacted only under microwave (Scheme 2, **7r**). We further investigated the impact of ring size on the reaction. The results showed that seven-membered fused ring was easily formed, while four-membered fused ring was more difficult to obtain (Scheme 2, **7s–7t**). In addition, fused dihydrophenanthridine derivatives were obtained in moderate yields using *o*-arylaniline-derived amides as substrates through the Morgan–Walls reaction [28] (Scheme 2, **7u–7v**).

Indole derived fused indolizidines and quinolizidines were also synthesized. Tryptamine-derived amides with indole nitrogen protected by methyl or benzyl were converted into corresponding products in good yields (Scheme 2, **7w–7x**). It was worth noting that unsubstituted indole derivatives led to rapid indolyl nitrogen sulfonylation by Tf<sub>2</sub>O, and these compounds were unsuitable for Movassaghi's method [20] to be activated and underwent B–N reaction. Under our reaction condition, indole derivatives without any substitution were converted into fused tetrahydro- $\beta$ -carboline with sulfonylations on indolyl nitrogen. The relatively low yields might be caused by insufficient Tf<sub>2</sub>O which was consumed by indolyl nitrogen sulfonylation. When Tf<sub>2</sub>O was adjusted to 2.2 equiv, the yields were improved obviously (Scheme 2, **7y–7z**).

Compounds **7o** and **7y** were converted into hydrochlorides, and their structures were unequivocally confirmed by X-ray crystal structure determination (Fig. 2).

Additionally, gram-scale syntheses of typical fused quinolizidines were also realized through the protocol (Scheme 3).

Our method was further applied to synthesize natural products or bioactive compounds. ( $\pm$ )-Crispine A (**7aa**, Scheme 4) shows cytotoxic activity against HeLa human cancer cell lines [29]. It was synthesized smoothly by using our one-pot method in good yield. A number of methods have been reported to synthesize ( $\pm$ )-crispine A, which undergoes multiple reaction steps in different ways [30–33], however, our protocol is one of the most efficient. Compound **7ab** (Scheme 4) is also a pharmacological active substance that displays potent antagonist activity at  $\alpha 4\beta 2$  nicotinic acetylcholine receptors (nAChR), a ligand-gated ion channel involved in a broad range of psychiatric and neurodegenerative disorders. In the reported method to synthesize compound **7ab**, highly corrosive POCl<sub>3</sub> and multi-step microwave-assisted

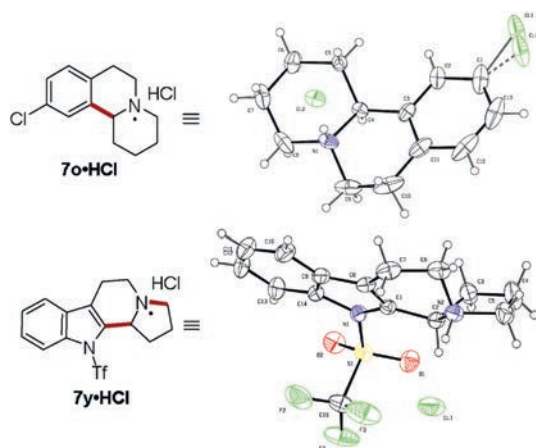


**Scheme 2.** Scope of this method. Uniform reaction conditions unless otherwise noted: Compound **6** (0.5 mmol),  $\text{TiF}_2\text{O}$  (1.1 equiv.), 2-F-Py (1.2 equiv.),  $\text{CH}_2\text{Cl}_2$  (5 mL), Ar,  $-78^\circ\text{C}$ , 30 min  $\rightarrow 40^\circ\text{C}$ , 2 h  $\rightarrow$  MeOH (5 mL),  $\text{NaBH}_4$  (2.0 equiv.), r.t., 2 h. <sup>a</sup>  $-78^\circ\text{C}$ , 30 min  $\rightarrow$  microwave,  $120^\circ\text{C}$ , 10 min  $\rightarrow$  r.t., 2 h. <sup>b</sup>  $\text{TiF}_2\text{O}$  (2.2 equiv.).

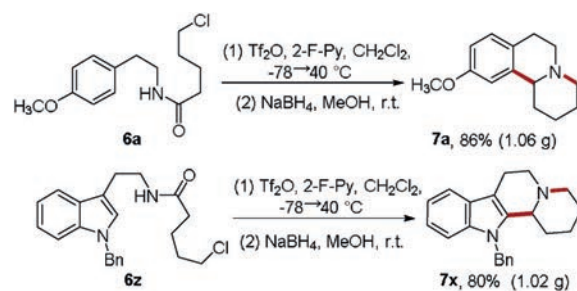
reaction with the temperature as high as  $150^\circ\text{C}$  were necessary [34]. Herein, synthesis was conducted in milder condition and more convenient method.

(±)-Harmicine (**7ac**, Scheme 4) exhibits strong antileishmania activity [35] and (±)-desbromoarborescidine A (**7ad**, Scheme 4) was an alkaloid discovered from the New Guinea tree *Dracontomelum mangiferum* [12]. These compounds were easily obtained from the detrifluoromethylsulfonylation of **7y** and **7z** in NaOH solution (Scheme 4).

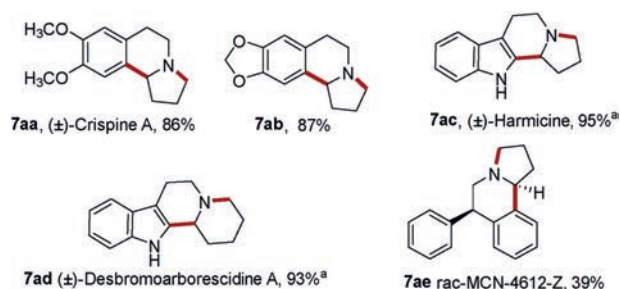
MCN-4612-Z (**7ae**, Scheme 4) is a potent uptake inhibitor for norepinephrine (NE), dopamine (DA) and serotonin (5-HT), which was developed as antidepressant drugs in 1980s [36]. It was synthesized using our protocol in 39% yield. And its



**Fig. 2.** X-ray crystal structure of **7o-HCl** and **7y-HCl**.



**Scheme 3.** Gram-scale synthesis.



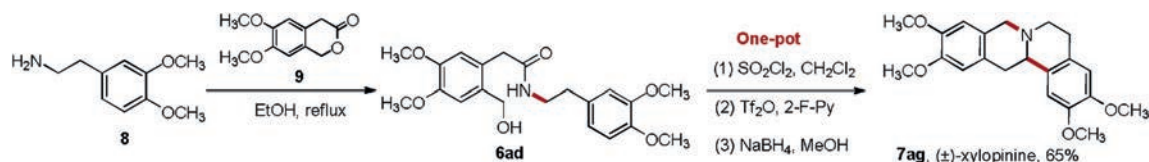
**Scheme 4.** Application of this method to the synthesis of natural products and bioactive compounds. Uniform reaction conditions unless otherwise noted: Compound **6** (0.5 mmol),  $\text{TiF}_2\text{O}$  (1.1 equiv.), 2-F-Py (1.2 equiv.),  $\text{CH}_2\text{Cl}_2$  (5 mL), Ar,  $-78^\circ\text{C}$ , 30 min  $\rightarrow 40^\circ\text{C}$ , 2 h  $\rightarrow$  MeOH (5 mL),  $\text{NaBH}_4$  (2.0 equiv.), r.t., 2 h. <sup>a</sup> Compound **7y** or **7z** (0.2 mmol), NaOH (4.0 mmol),  $\text{H}_2\text{O}$  (4 mL), MeOH (16 mL), reflux, 2 h.

diastereoisomer (**7af**, MCN-4612-E, Supporting information) [37] was also isolated in the same reaction in 45% yield.

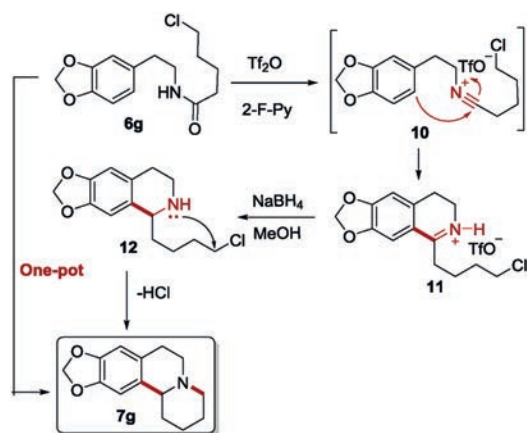
Furthermore, a new strategy was designed to synthesize protoberberine (±)-xylopinine (**7ag**, Scheme 5). (±)-Xylopinine is an alkaloid found in the stem of *Xylopia laevigata* (Annonaceae), which exhibits strong *in vitro* cytotoxic activity toward various tumor cell lines [38]. The amide substrate (**6ad**, Scheme 5) was synthesized from 3,4-dimethoxyphenylethylamine (**8**, Scheme 5) and 6,7-dimethoxyisochroman-3-one (**9**, Scheme 5), which was then subjected to chlorination, B-N reaction, reduction of Schiff base intermediate, and intramolecular cyclization in one-pot to offer (±)-xylopinine as a racemate in 65% total yield. Among the reported synthesis strategies for tetrahydroberberines [39–41], our method is the one of the most efficient.

A plausible mechanism is illustrated in Scheme 6 taking compound **7g** as an example. Amide substrate (**6g**, Scheme 6) was firstly activated by  $\text{TiF}_2\text{O}$  to obtain a nitrilium ion [42,43] (**10**, Scheme 6), which underwent Bischler-Napieralski reaction to obtain cyclic iminium ion (**11**, Scheme 6). The iminium ion **11** was isolated and its structure was confirmed by  $^1\text{H}$  NMR,  $^{13}\text{C}$  NMR and HRMS. **11** was subjected to reduction of the C=N bond (**12**, Scheme 6) followed by intramolecular nucleophilic substitution to obtain polycyclic product **7g**.

In summary, we propose a facile tandem protocol to construct fused indolizidines and quinolizidines. The protocol is efficient because it integrates three step reactions in one-pot and does not require isolation and purification of any intermediate. The protocol can be easily scaled up to gram-scale. Taking advantage of the mild reaction conditions and functional group tolerance, the protocol was successfully applied to synthesize natural products such as (±)-crispine A, (±)-desbromoarborescidine A, (±)-harmicine, (±)-xylopinine and other bioactive substances.



Scheme 5. Total synthesis of protoberberine (±)-xylopinine.



Scheme 6. The proposed mechanism.

### Declaration of competing interest

The authors report no declarations of interest.

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### Appendix A. Supplementary data

Supplementary material related to this article can be found, in the online version, at doi:<https://doi.org/10.1016/j.ccl.2021.01.018>.

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